#### **Lipid Biosynthesis**



Craig Wheelock February 2nd, 2009

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#### Why do we care about lipids?

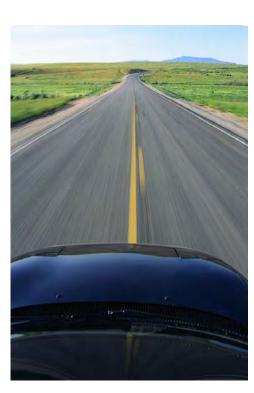
- ~80% of European population overweight
- ~ 1/3 obese
  - >=130 million obese adults in EU
  - >6% of total health costs
  - >10-13% of deaths in Europe

Contributes to diabetes, coronary heart disease, hypertension, stroke & cancer

Diseases of dyslipidemia are one of the greatest health challenges of the 21st century

#### **Outline**

- Lipid overview: synthesis and structure
- Fatty acids
- Eicosanoids
- Break . . . .
- Triacylglycerols
- Phospholipids
- Cholesterol
- Bile acids, enterohepatic circulation
- Summary



#### What is a lipid?

#### **Classical definition:**

biological molecule that is soluble in organic solvent, but insoluble in water

#### Modern (specific) definition:

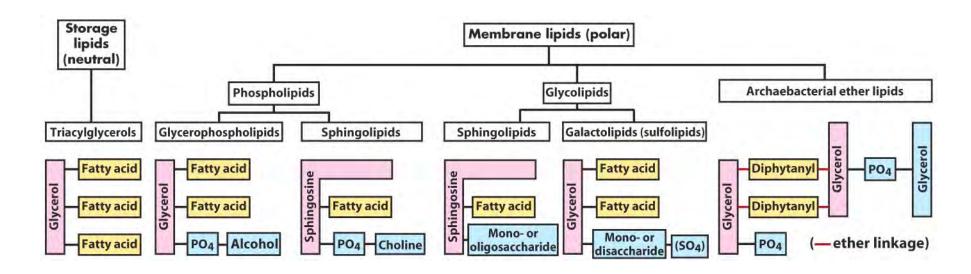
fatty acids and their derivatives, and substances related biosynthetically or functionally to these compounds

#### **More specific definition:**

Hydrophobic small molecules that originate by carbanionbased condensations of thioesters (fatty acids, polyketides, etc.) and/or by carbocation-based condensations of isoprene units (prenols, sterols, etc)

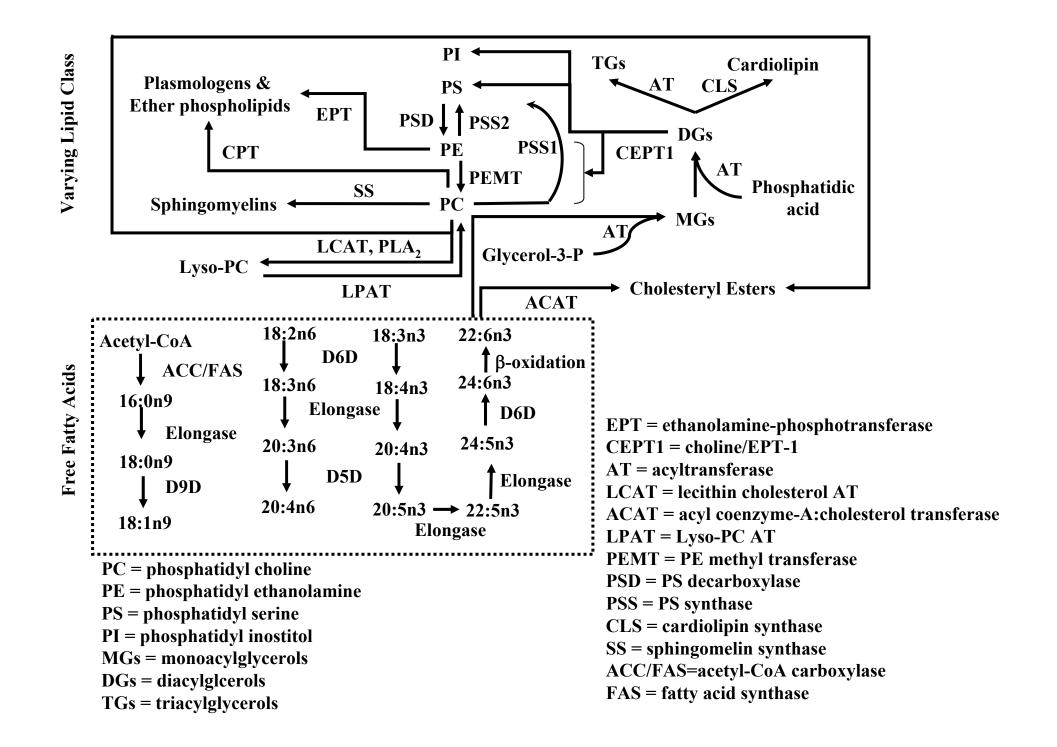
#### Vary greatly in structure and function

## Some typical lipids

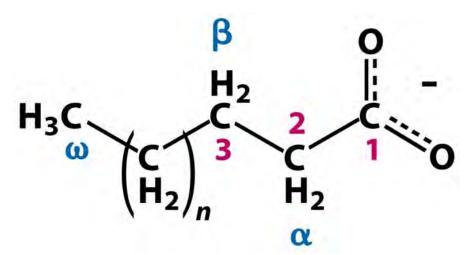


#### In addition:

cholesterol, cofactors, vitamins, bile acids, hormones, eicosanoids, other signaling molecules, etc......



# Fatty acids (FA)



- compounds synthesized in nature via condensation of malonyl coenzyme A units by a fatty acid synthase complex
- contain even numbers of carbon atoms in straight chains (commonly C14- C24)
- may be saturated or unsaturated
- can contain variety of substituent groups

#### **Fatty Acids**

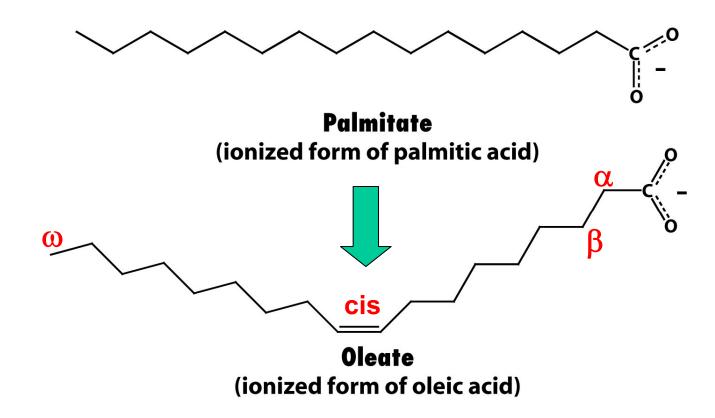
• Single(σ)-bonded carbon chains with a terminal carboxylic acid:



- Produced de novo by animals and plants
- Can contain double( $\pi$ )-bonded carbons which are inserted by desaturases

#### **Unsaturation & shorter chain**

- ↓ melting point
- † membrane fluidity (cis bond gives the "kink" in the carbon chain)

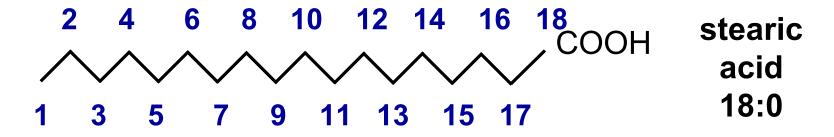


 Fatty acids are sigma-bonded carbon chains with a carboxylic acid functional group



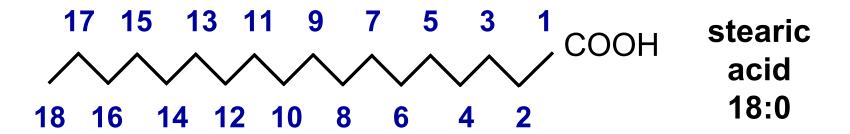
 Individual fatty acids can be identified by one of two numerical nomenclature systems

n-Designation
Carbon numbering starts from methyl end

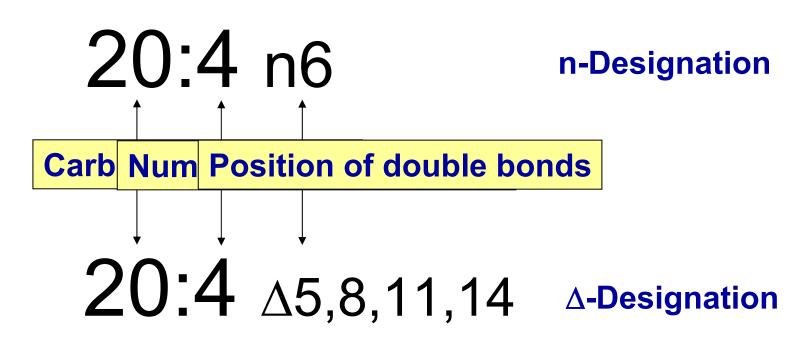


 Individual fatty acids can be identified by one of two numerical nomenclature systems

**∆-Designation Carbon numbering starts from carboxyl group** 

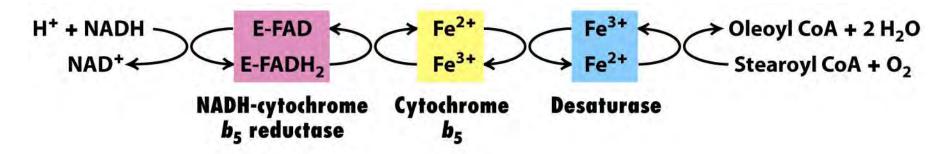


Standard nomenclature (arachidonic acid):

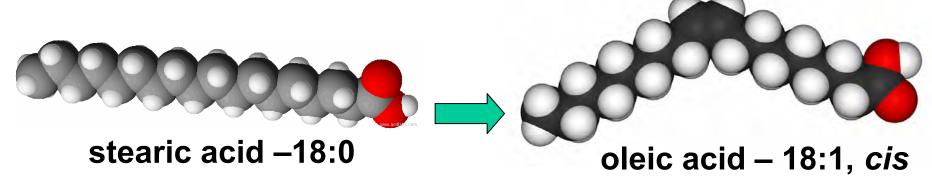


## **Desaturation of Fatty Acids**

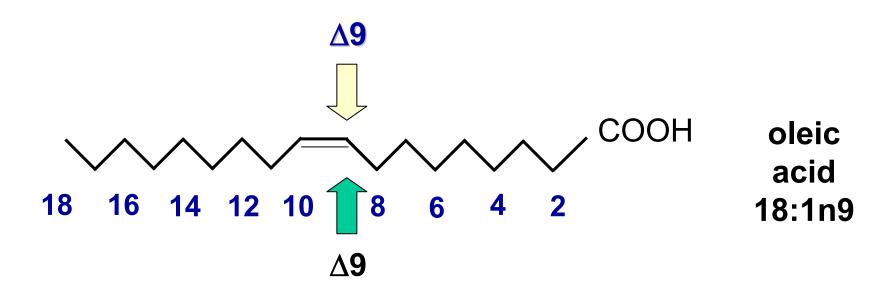
complex of 3 membrane proteins in E.R.



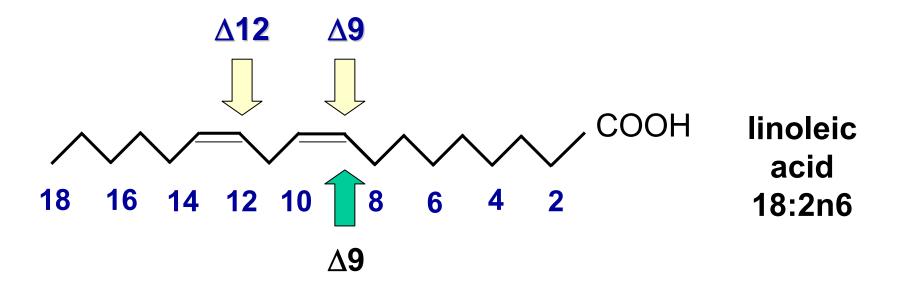
series of desaturase enzymes creates positionspecific double bonds



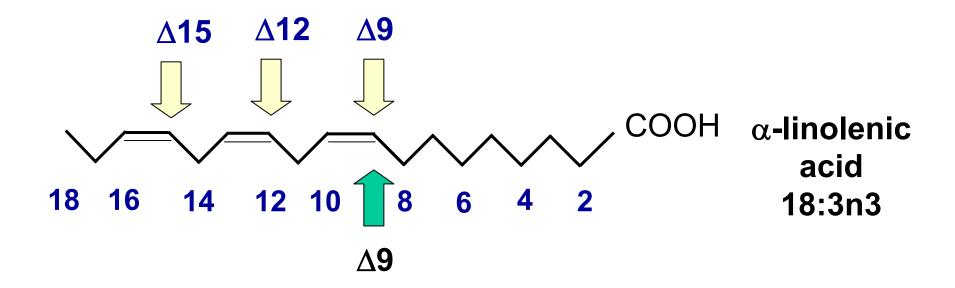
 The first desaturation of a saturated fatty acid is always at the ∆9 position



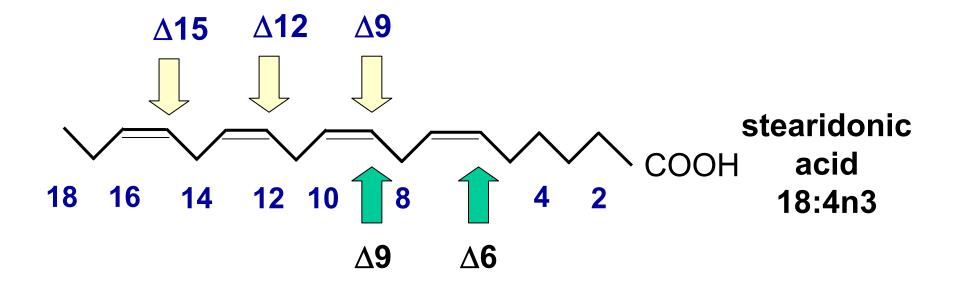
Plants can also desaturate at the Δ12



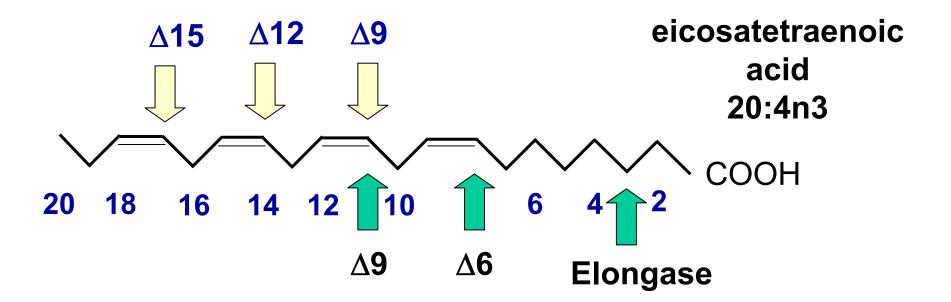
 Plants can also desaturate at the ∆12 and the ∆15 carbon



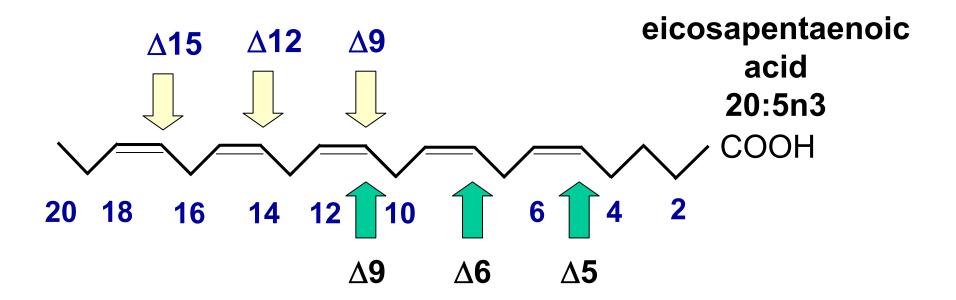
 Animals desaturate plant-derived poly unsaturated fatty acids (PUFAs) at the ∆6 carbon



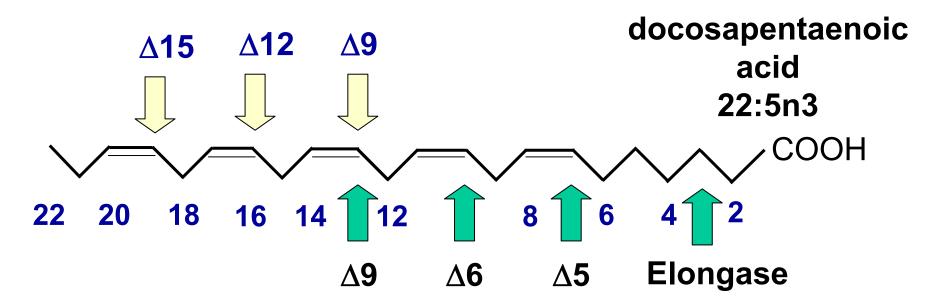
 To add another double bond, animals must first elongate the fatty acid



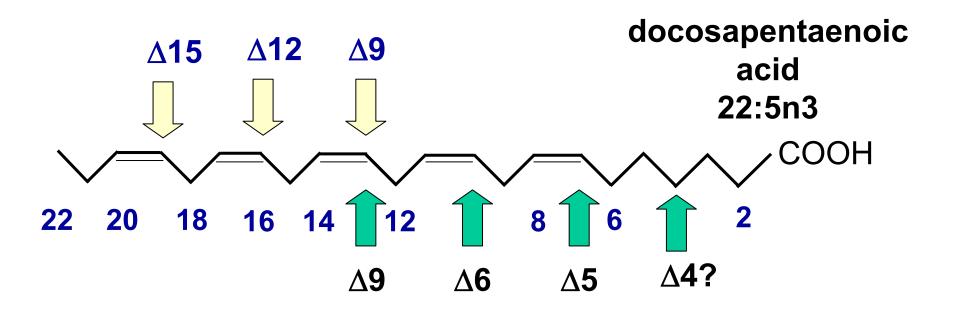
Animals can then add a ∆5 double bond



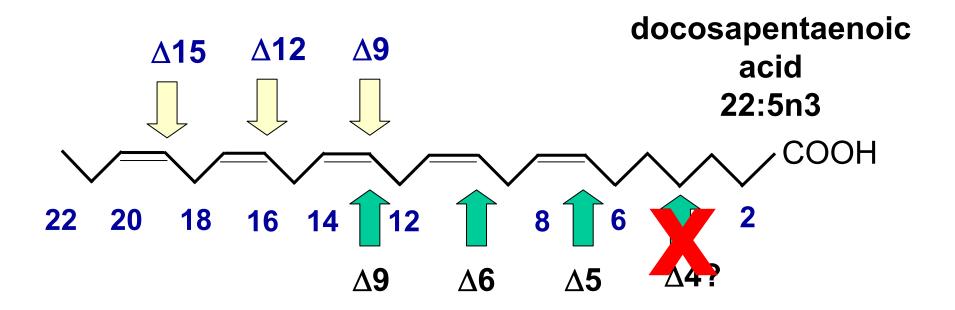
 To make docosahexaenoic acid (DHA, 22:6n3) animals must further elongate the acyl chain to 22 carbons



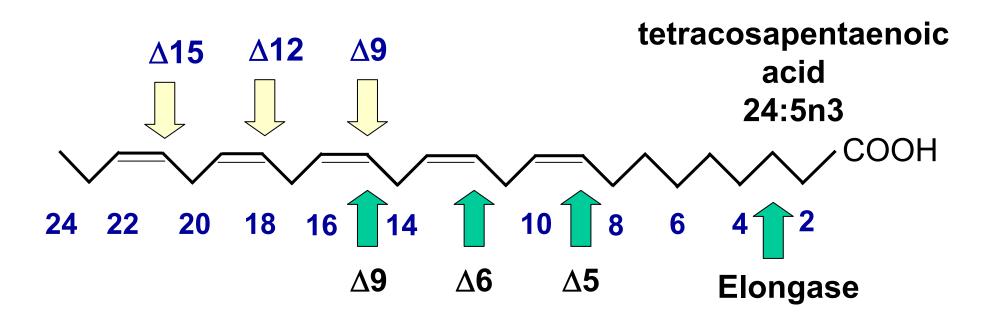
 Another double bond is inserted at the ∆4 position



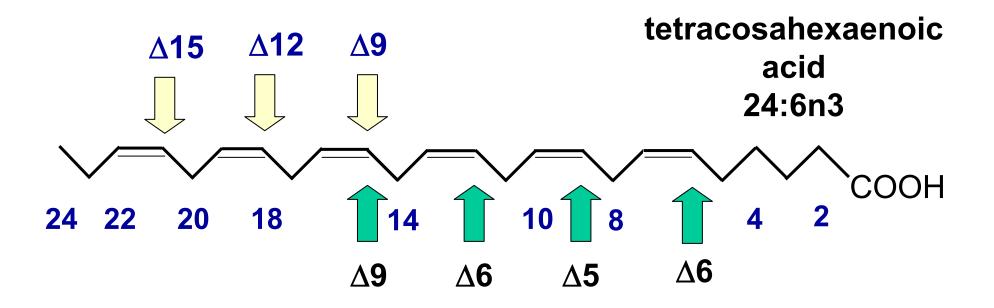
BUT THERE IS NO ∆4 DESATURASE



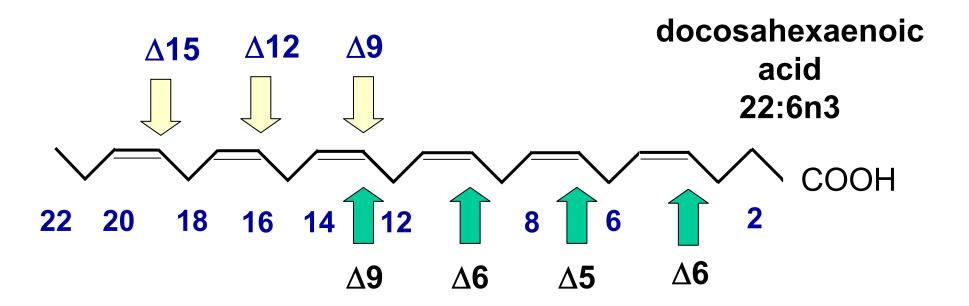
 To make DHA (22:6n3) the fatty acid must first be elongated again to a 24 carbon chain



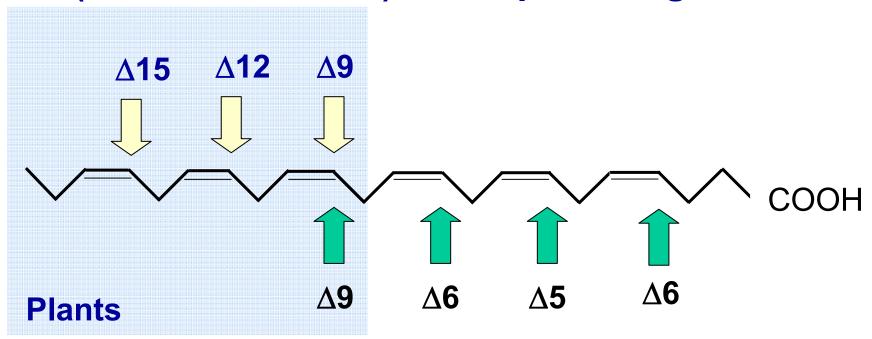
 Then the chain can be acted upon again by the ∆6 desaturase yielding 24:6n3



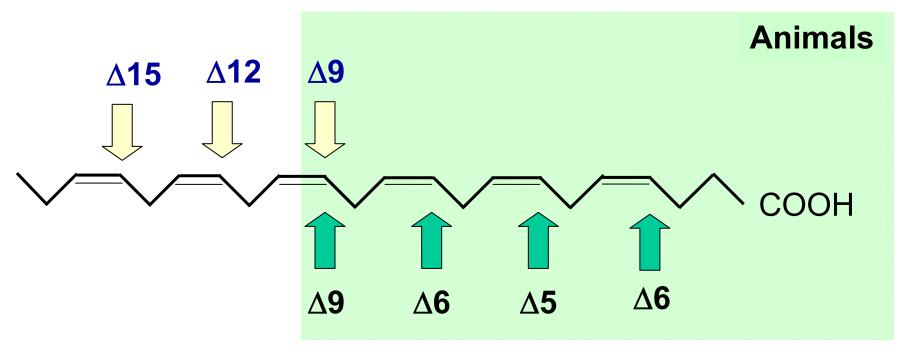
 The last step in DHA synthesis is a 2-carbon chain shortening by peroxisomal β-oxidation



 Fatty acids with double bonds on the methylene side of an original ∆9 double bond (n3 and n6 bonds) are of plant origin



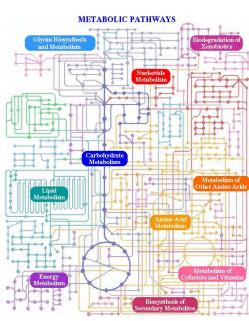
 Animals then modify these polyunsaturated fatty acids utilizing their own distinct set of desaturases.



- Why two designations?
  - One is useful to describe biochemical reactions
  - One is useful to track families of fatty acids in nutrition

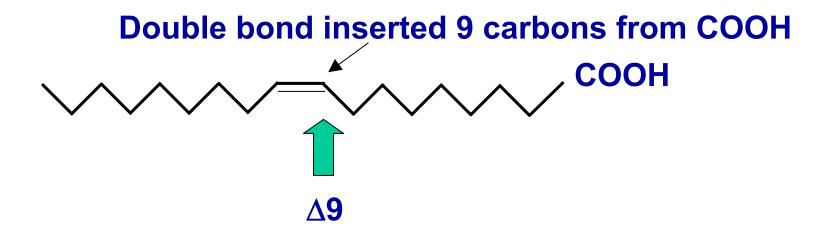


VS



#### **Biochemical Reactions**

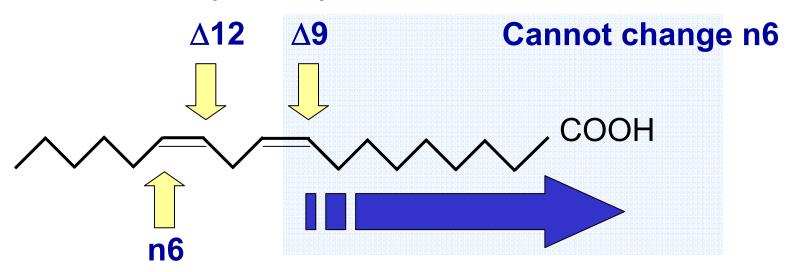
- Desaturases and elongases act from the carboxy-terminus of the fatty acid
- Therefore, the ∆-designation is useful to describe the biochemistry of fatty acid metabolism



#### **Nutrition**

 The n-designation is useful as it allows nutritionists to link diet with tissue fatty acid composition

Linoleic acid (18:2n6)



## Omega fatty acids (ω)

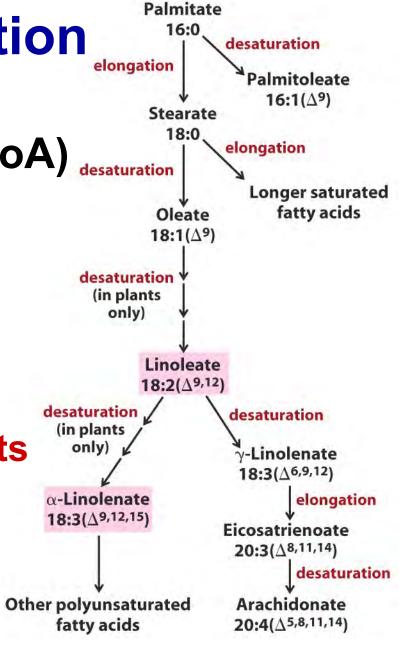
- Nomenclature based upon position of the first double bond relative to the carbon chain terminal methyl
- Omega-3 fatty acids cannot be synthesized de novo by humans, obtained from fish

Omega-6 fatty acids are obtained from diet (grains, etc) and can be synthesized

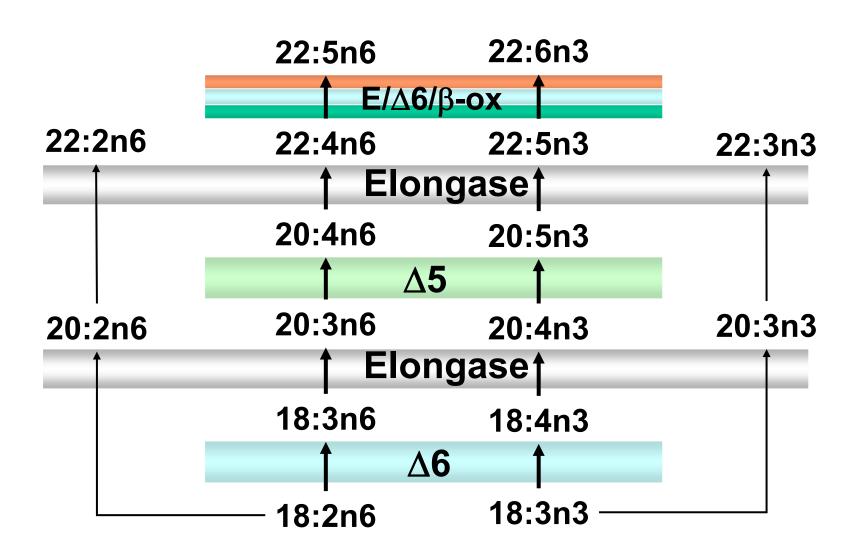
arachidonic acid (AA)

#### Fatty acid modification

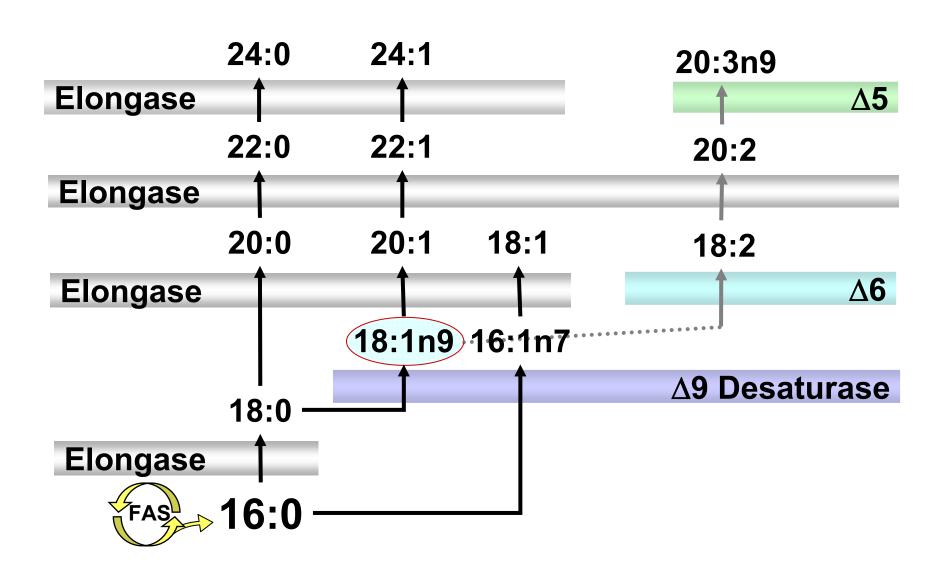
- Activated FFA (acyl-CoA)
- Elongation
- β-oxidation
- Desaturation
  - up to C9 in animals
- >C9 occurs only in plants



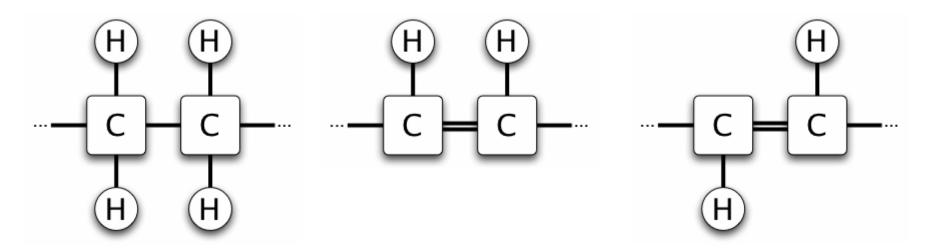
#### **Essential FA Metabolism**



#### De Novo Metabolism



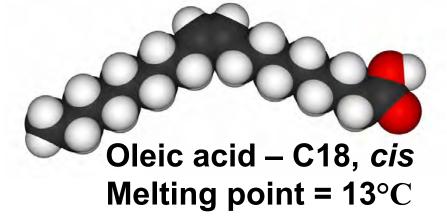
## Trans-fat contains "trans" vs. "cis" bond

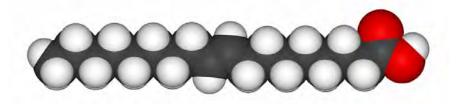


saturated

cis double bond

trans double bond





Elaidic acid – C18, *trans*Melting point = 45°C

TABLE 12.1 Some naturally occurring fatty acids in animals

Number of carbons	Number of double bonds	Common name	Systematic name	Formula
12	0	Laurate	n-Dodecanoate	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>10</sub> COO-
14	0	Myristate	n-Tetradecanoate	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>12</sub> COO-
16	0	Palmitate	n-Hexadecanoate	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>14</sub> COO
18	0	Stearate	n-Octadecanoate	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>16</sub> COO
20	0	Arachidate	n-Eicosanoate	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>18</sub> COO-
22	0	Behenate	n-Docosanoate	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>20</sub> COO-
24	0	Lignocerate	n-Tetracosanoate	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>22</sub> COO-
16	1	<b>Palmitoleate</b>	cis-∆9-Hexadecenoate	CH, (CH,),CH=CH(CH,),COO-
18	(1)	Oleate	cis-∆9-Octadecenoate	CH, (CH,),CH=CH(CH,),COO
18	2	Linoleate	$cis$ , $cis$ - $\Delta^9$ , $\Delta^{12}$ - $Octadecadienoate$	$CH_3$ $(CH_2)_4$ $(CH = CHCH_2)_2$ $(CH)_6$ $COO^{-1}$
18	3	Linolenate	all-cis- $\Delta^9$ , $\Delta^{12}$ , $\Delta^{15}$ -Octadecatrienoate	$CH_3CH_2(CH=CHCH_2)_3(CH_2)_6COO^-$
20	4	Arachidonate	all- <i>cis</i> $\Delta^5$ , $\Delta^8$ , $\Delta^{11}$ , - $\Delta^{14}$ Eicosatetraenoate	$CH_3(CH_2)_4(CH = CHCH_2)_4(CH_2)_2COO$

Ex) Linoleate, linoleic acid 18:2n6 cis,cis- $\Delta^9$ , $\Delta^{12}$  –octadecatrienoate an  $\omega$ -6 fatty acid

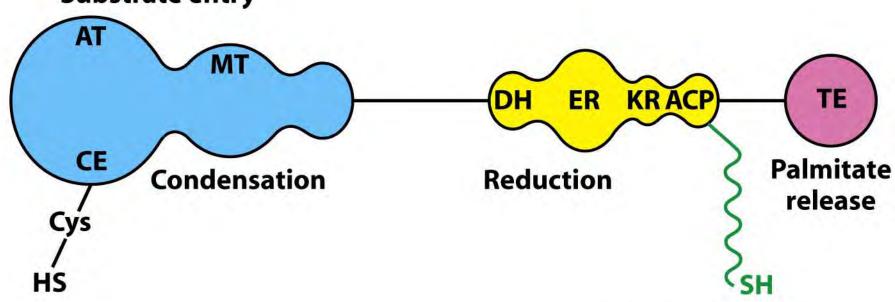
## Fatty acid biosynthesis

- Where? In the cytoplasm in: liver, adipose and mammary glands
- How? Stepwise incorporation of 2 carbon atoms from Acetyl-CoA
- •Cost? both ATP and NADPH

# Overall equation for synthesis of 16:0 (palmitate, hexadecanoate)

## Fatty acid synthase system

**Substrate entry** 



**Translocation** 

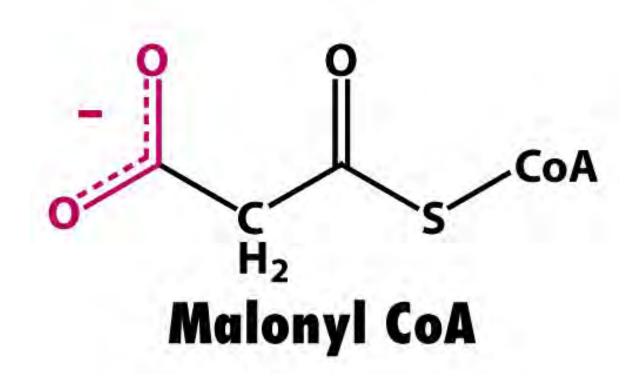
Domain 3:

#### **Domain 1:**

AT acetyl transferase
MT malonyl transferase
CE condensing enzyme
(=β-ketoacyl synthase, KS)

#### Domain 2:

DH dehydratase TE thioesterase ER enoyl reductase KR β-ketoacyl reductase ACP acyl carrier protein

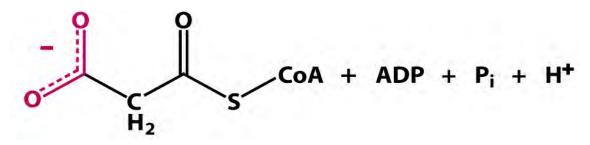


"activated 2-carbon donor" committed step in FA synthesis

## Coenzyme A (CoA-SH)

#### **Acetyl-CoA**

# Acetyl-CoA-carboxylase (ACC) the committed step in FA synthesis



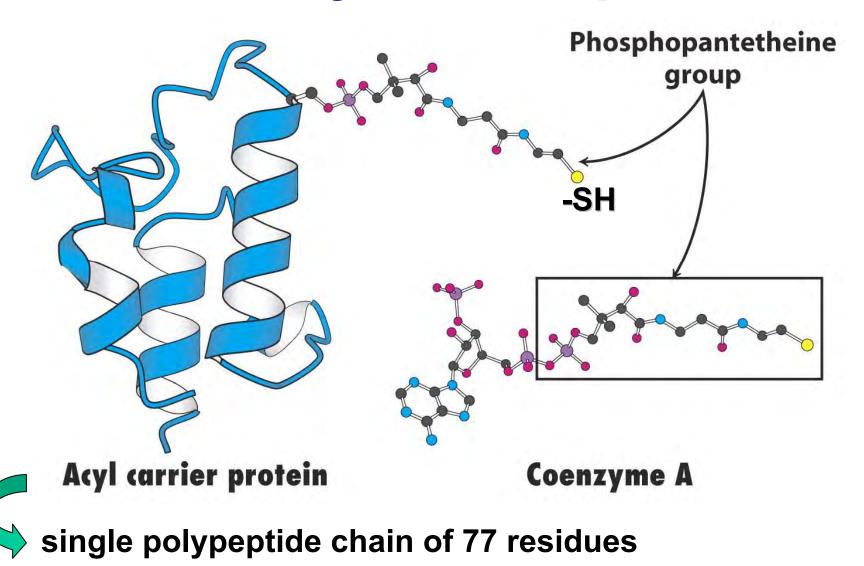
Malonyl-CoA

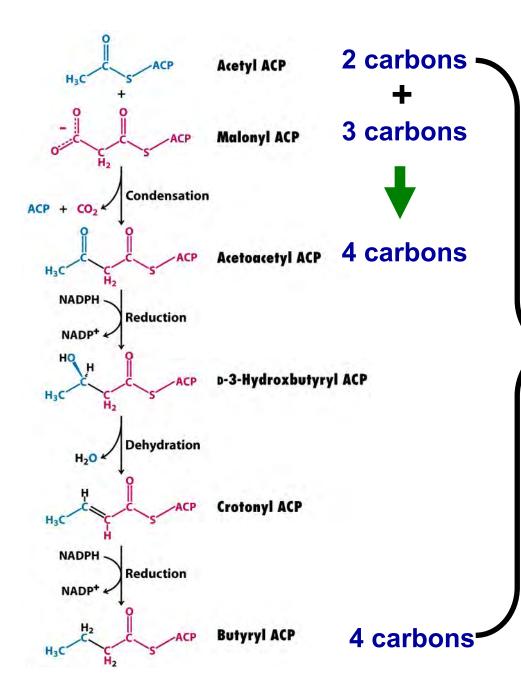
# Elongation phase of FA synthesis

Acetyl transacylase
Acetyl CoA + ACP ↔ acetyl ACP + CoA

Malonyl transacylase
Malonyl CoA + ACP ↔ malonyl ACP + CoA

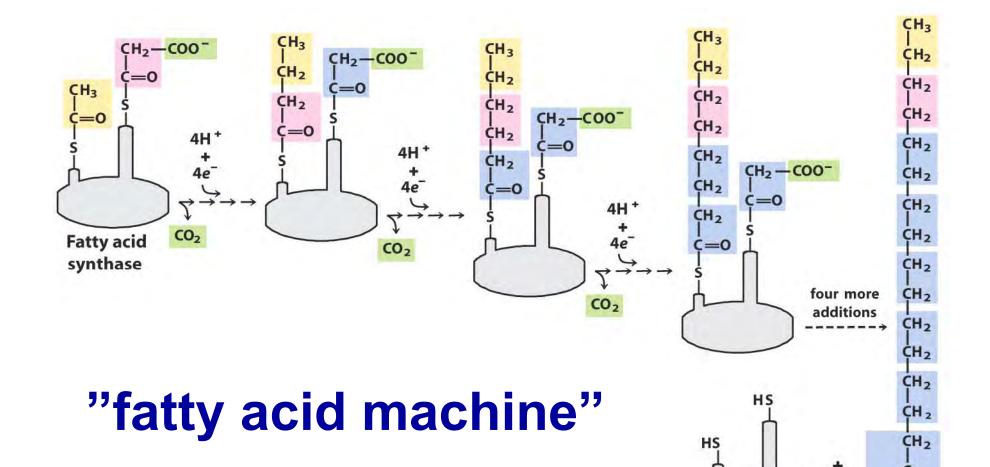
## ACP – acyl carrier protein





# Fatty acid synthase reaction sequence

7 rounds gives: 16:0, palmitate



**Palmitate** 

**AT** = acetyl transferase

**MT** = malonyl transferase

**KS** =  $\beta$ -ketoacyl synthase,

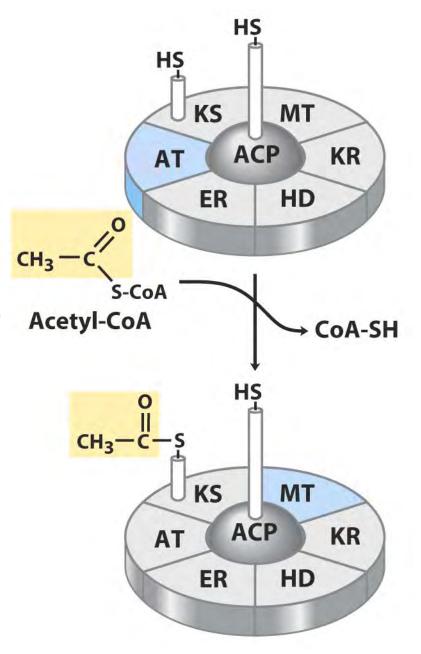
(CE condensing enzyme)

**HD** = dehydratase

**ER** = enoyl reductase

**KR** =  $\beta$ -ketoacyl reductase

ACP = acyl carrier protein



**AT** = acetyl transferase

MT = malonyl transferase

**KS** =  $\beta$ -ketoacyl synthase,

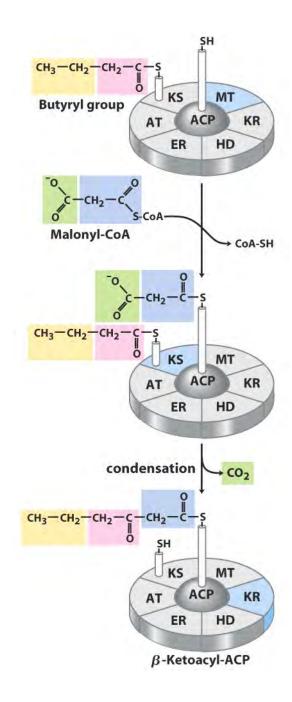
(CE condensing enzyme)

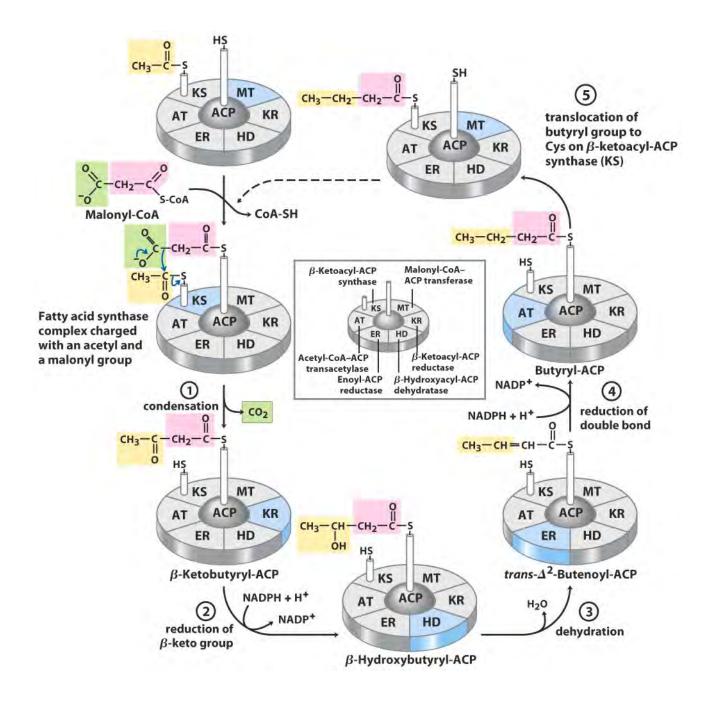
**HD** = dehydratase

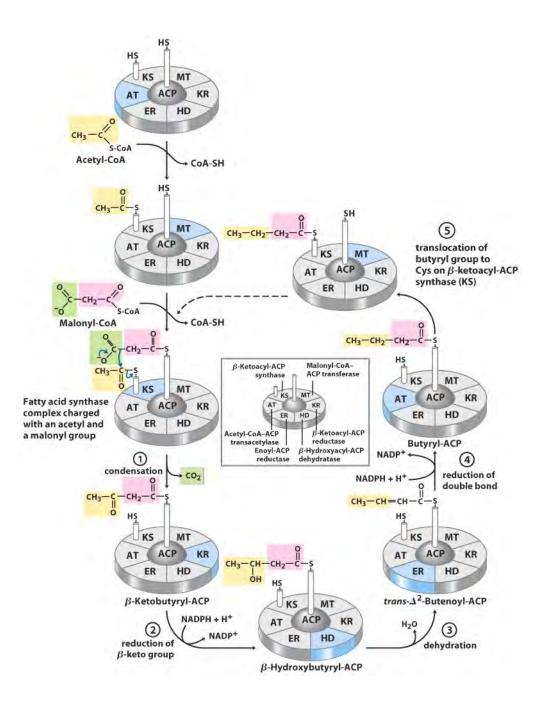
**ER** = enoyl reductase

 $KR = \beta$ -ketoacyl reductase

**ACP** = acyl carrier protein

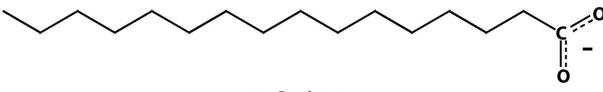






# Ex) The complete reaction for synthesis of 16:0

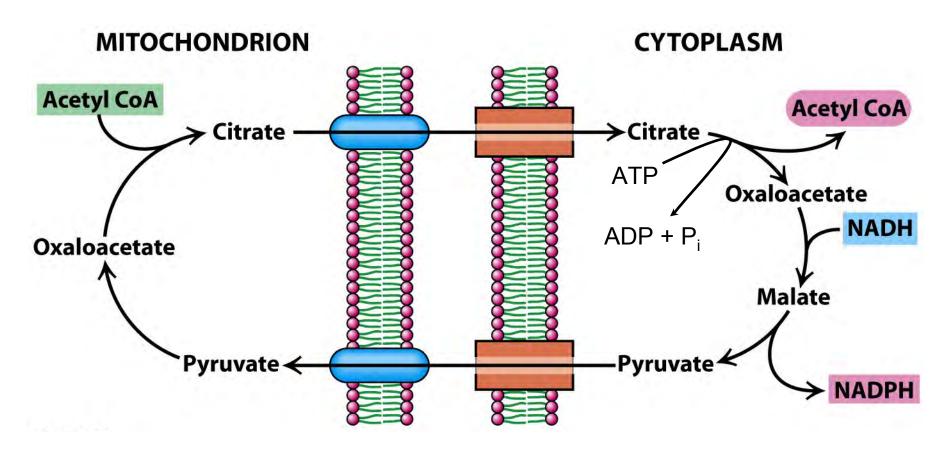
$$16:0 + 8 CoA + 7ADP + 7P_i + 14NADP^+ + 6H_2O$$



**Palmitate** (ionized form of palmitic acid)

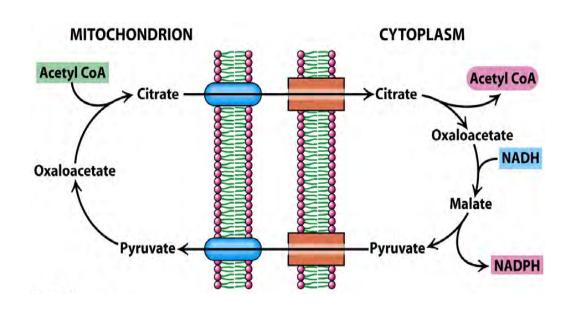
#### Transfer of Acetyl-CoA to the cytosol

→FA synthesized in cytoplasm, acetyl CoA formed from pyruvate in mitochondria



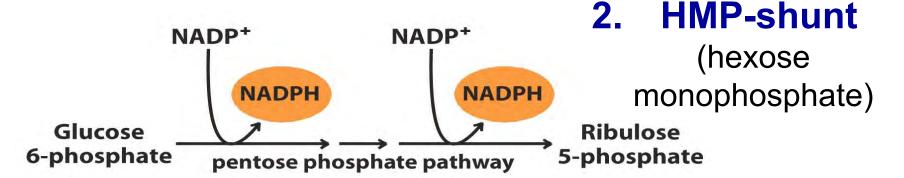
Citrate + ATP + CoA + $H_2O \rightarrow$  acetyl CoA + ADP +  $P_i$  + oxaloacetate

## 2 Ways to Generate NADPH



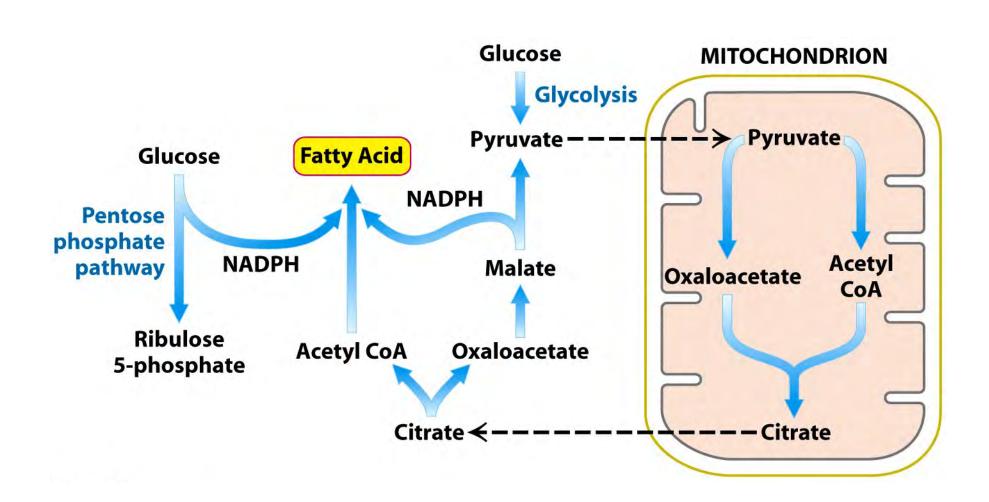
#### 1. Malic enzyme

(NADP+-linked malate enzyme)

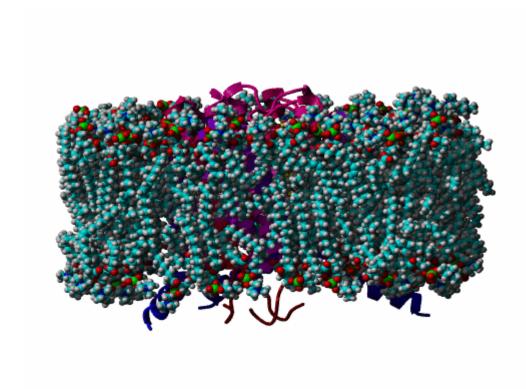


occurs exclusively in the cytoplasm = accounts for 60% of NADPH

## Fatty acid synthesis requires the integration of multiple metabolic pathways



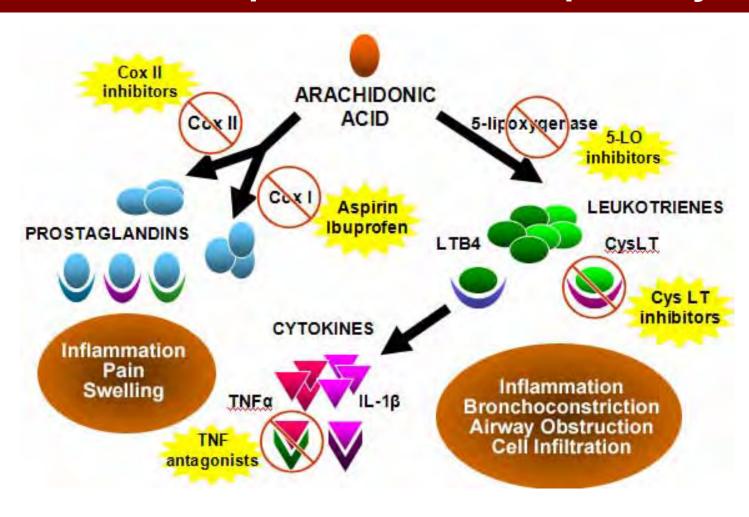
### **EICOSANOIDS**





#### AA is metabolized to inflammatory mediators

## Many current anti-inflammatory & pain medicines inhibit some portion of the AA pathway



# Eicosanoid hormones are derived from PUFAs

 Arachidonic acid (20:4n6) is major precursor of multiple signal molecules:

prostaglandins (PG), prostacyclins, thromboxanes (TX) and leukotrienes (LT)

- PGs = 20 carbon fatty acids containing a 5carbon ring
- PGs stimulate inflammation, regulate blood flow, control ion transport, modulate synaptic transmission & induce sleep

COO-CH<sub>3</sub>

Thromboxane  $A_2$  (TXA<sub>2</sub>)

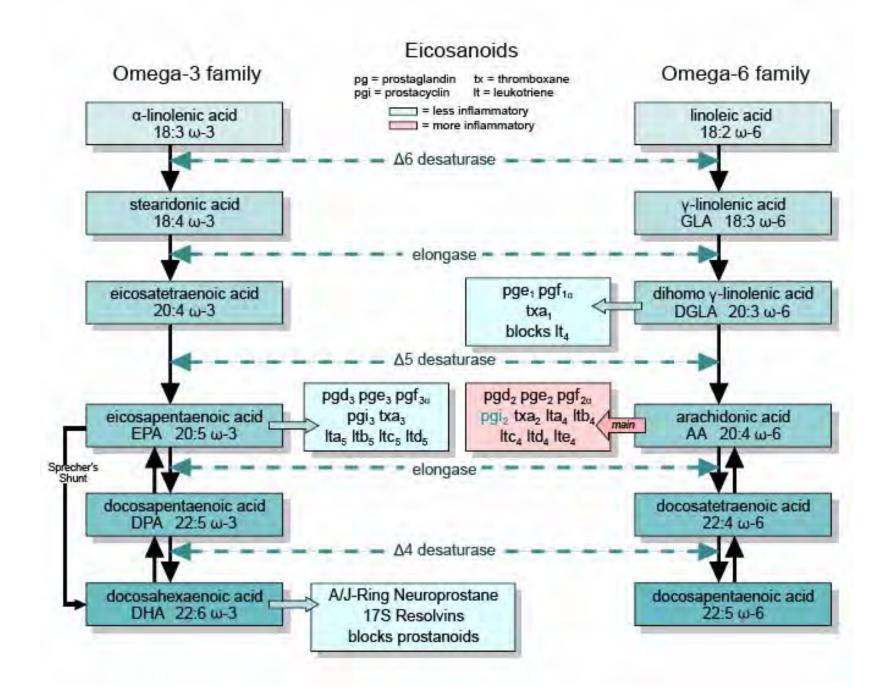
Leukotriene B<sub>4</sub>

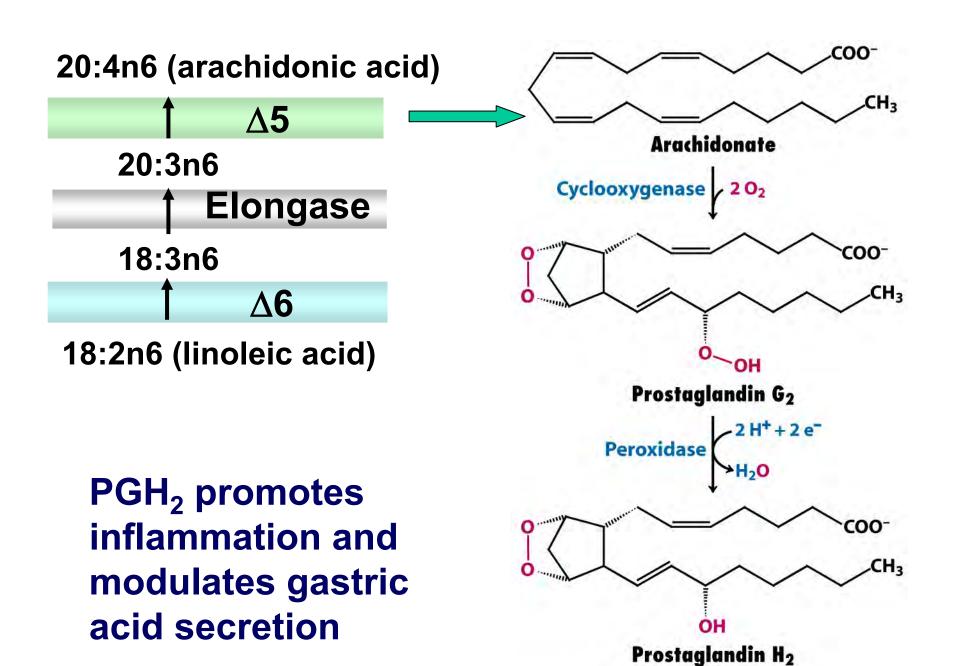
#### Importance of eicosanoids

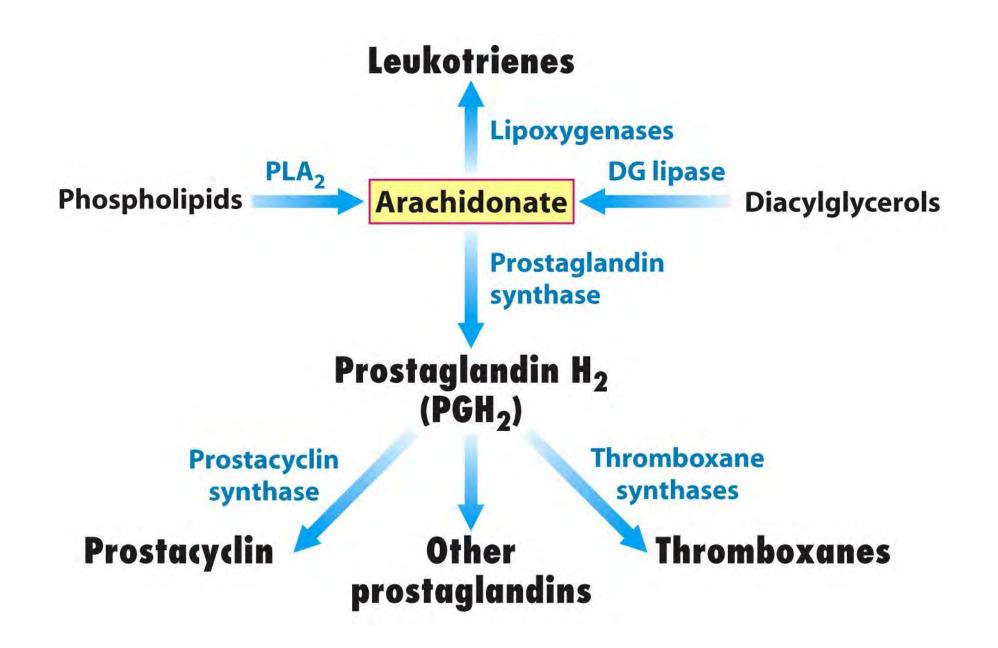
- Nobel Prize in 1982 for discovery of PG biological role and 1990 for PG synthesis
- PGs are found in almost all tissues & organs
- Ex of PG antagonists:
  - NSAIDs (inhibit COX → aspirin, ibuprofen)
  - corticosteroids (inhibit phospholipase A2 production)
- TXs are vasoconstrictors & hypertensive agent
  - role in thrombosis (clot in blood vessel)
- LTs & cysteinyl-LTs important in inflammation
  - asthma, psoriasis, anaphylaxis & atherosclerosis

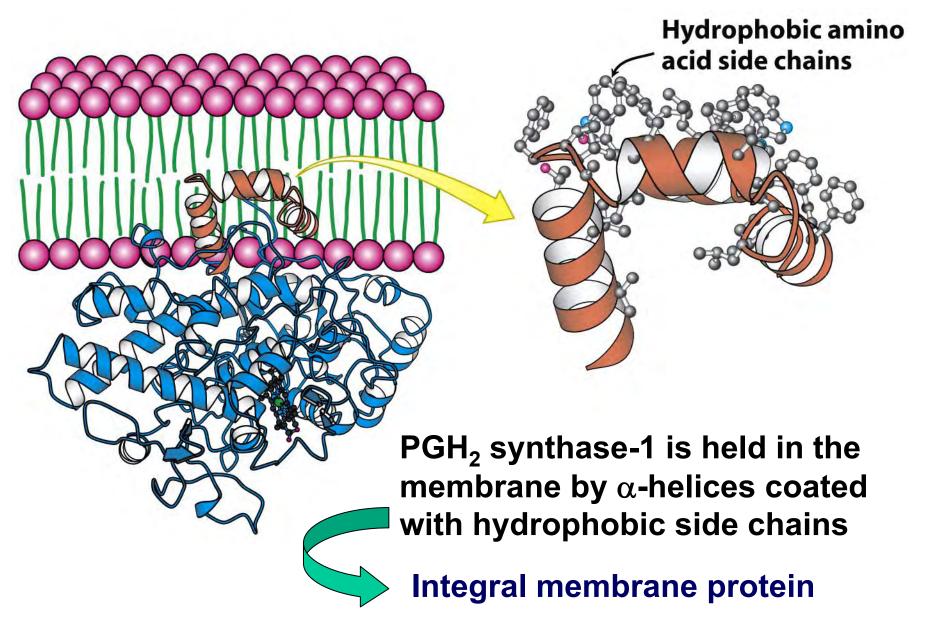
#### **Eicosanoid-based medicines**

Type	Medical condition		
PGI <sub>1</sub> analog	Pulmonary hypertension, avoiding reperfusion injury		
PG analog	Glaucoma, ocular hypertension		
PG analog	Labor induction		
PGE <sub>2</sub>	Labor induction		
PGI <sub>2</sub> analog	Pulmonary arterial hypertension		
PGE <sub>1</sub> analog	Stomach ulcers, labor induction		
LT receptor antagonist	Asthma, seasonal allergies		
PGI analog	Pulmonary hypertension		

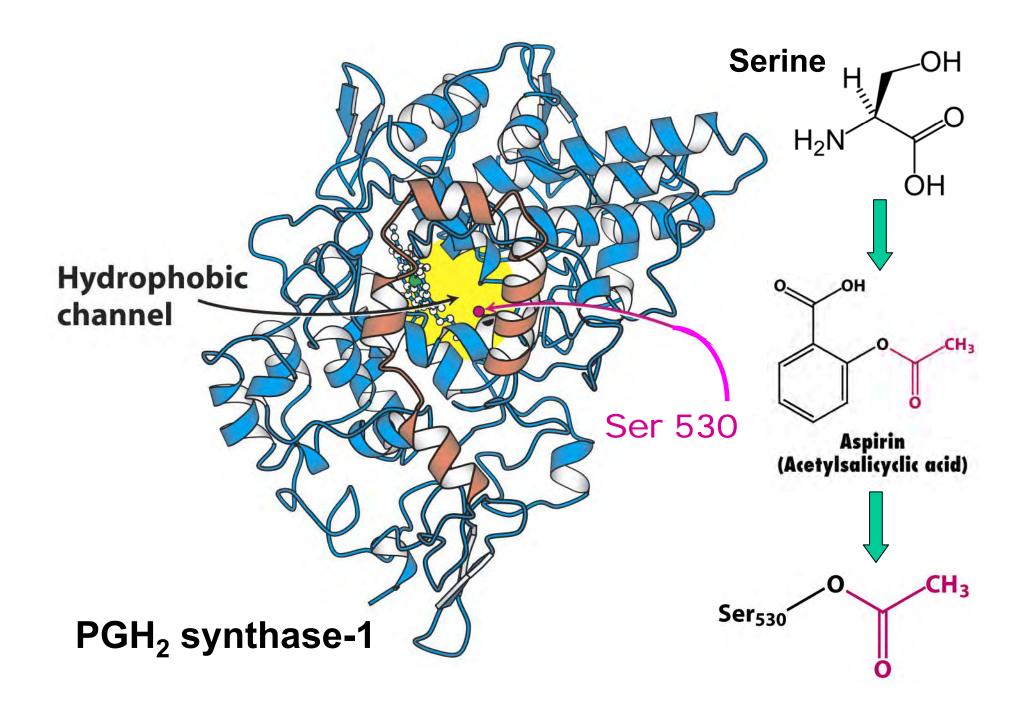




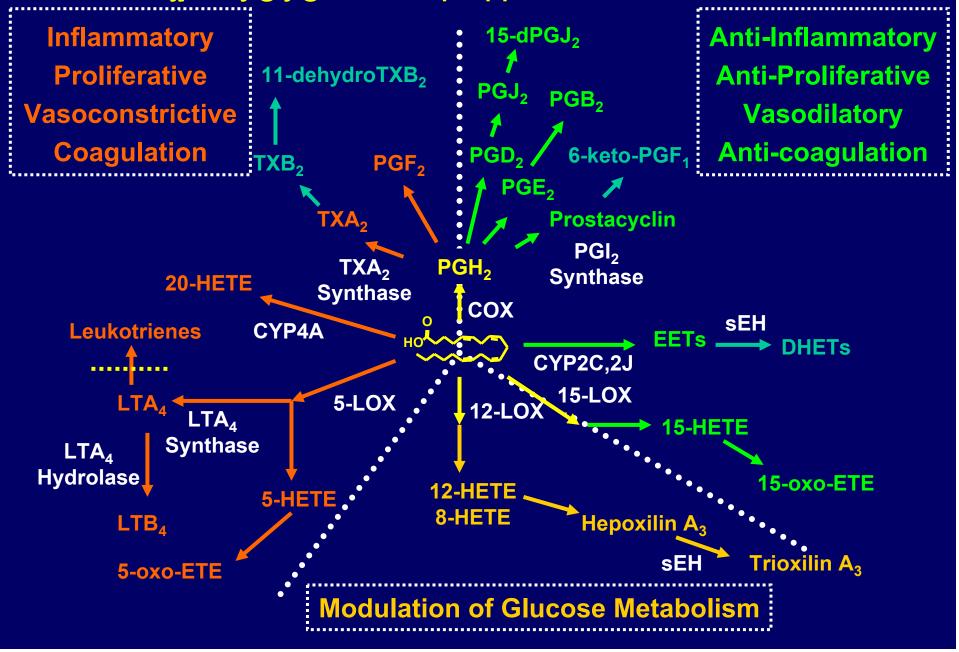




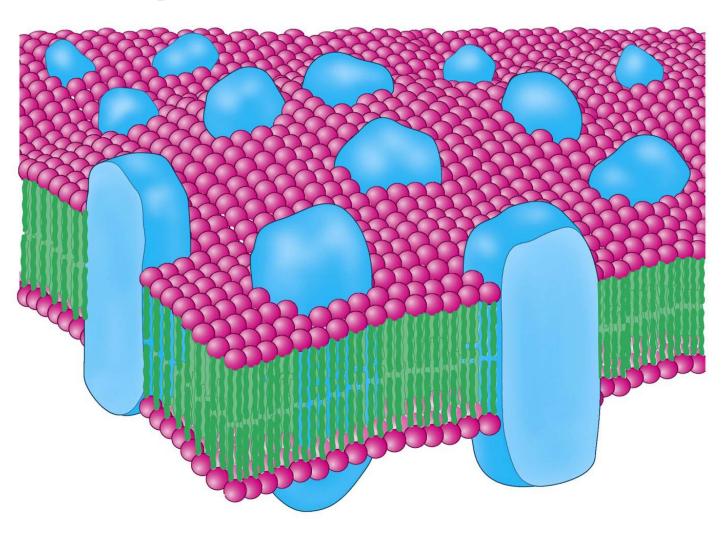
AA is hydrophobic and is funneled thru protein channel

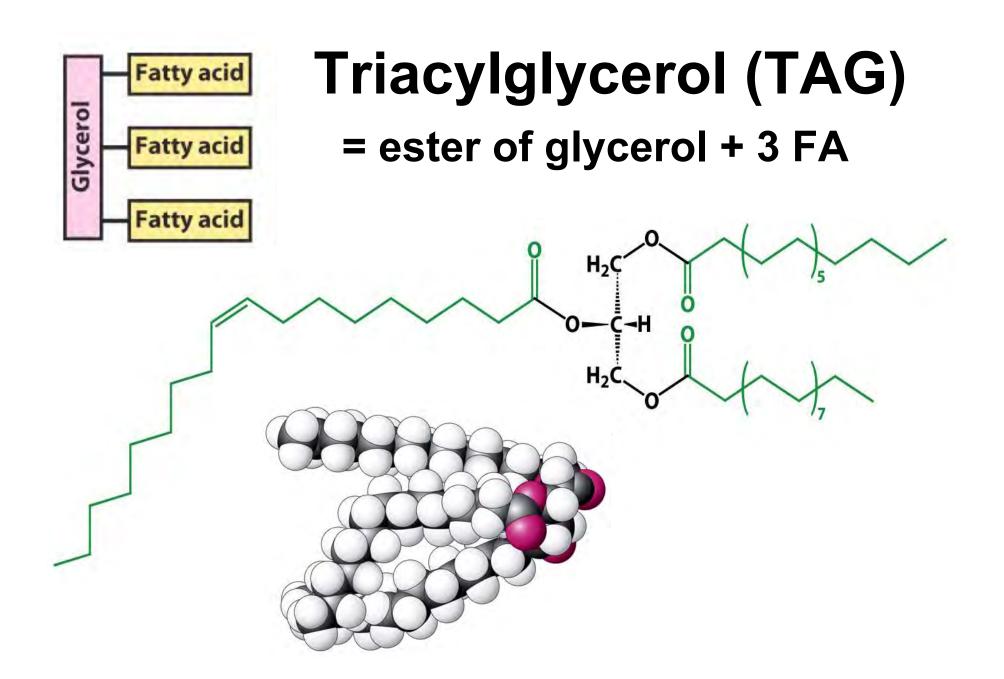


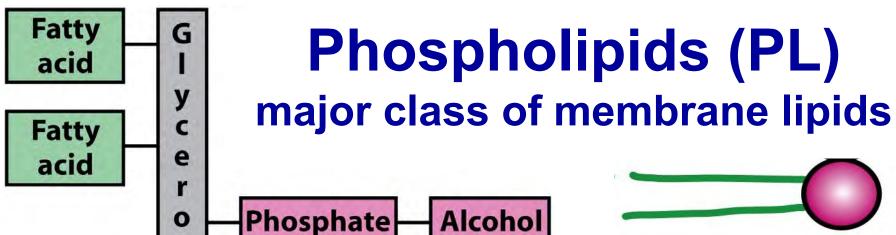
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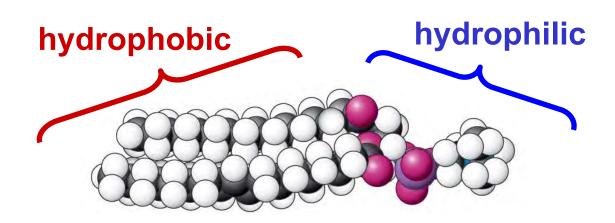
# Biosynthesis of membrane lipids and steroids

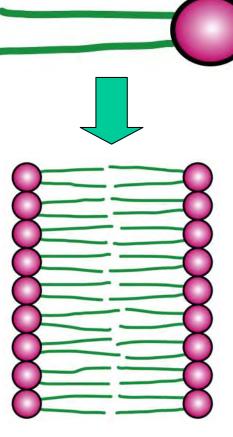






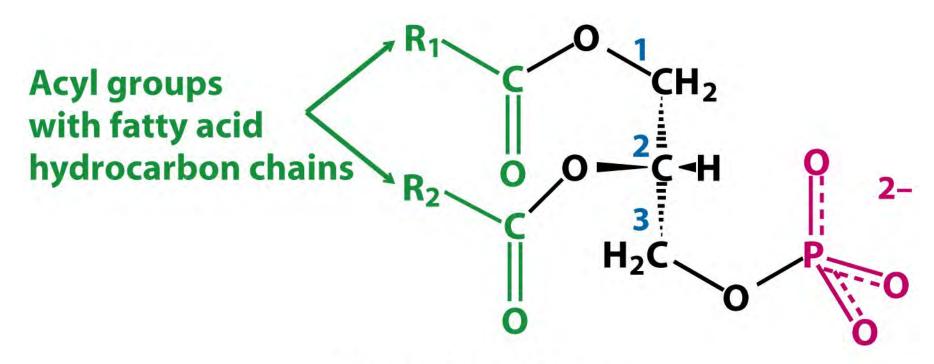
Glycerophospholipid constructed of 4 components





**Membrane** 

# 1<sup>st</sup> step in synthesis of phospholipids (PLs) and triacylglycerols (TAGs) is the synthesis of phosphatidate



Phosphatidate (Diacylglycerol 3-phosphate)

HO
$$CH_2$$
 $HO$ 
 $CH_2$ 
 $H_2C$ 
 $O$ 
 $CO$ 
 $COA$ 
 $CO$ 

**Glycerol 3-phosphate** 

Lysophosphatidate

Phosphatidate — is built from L-glycerol-3-phosphate and activated fatty acids

Usually saturated

R<sub>2</sub>CO-CoA CoA

R<sub>2</sub>CO-CoA CoA

R<sub>2</sub>CO-CoA CoA

R<sub>2</sub>CO-CoA CoA

R<sub>2</sub>CO-CoA CoA

R<sub>2</sub>CO-CoA CoA

Usually unsaturated

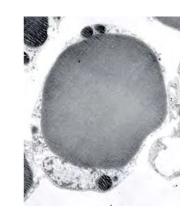
**Phosphatidate** 

## TAG synthesis proceeds via DAG

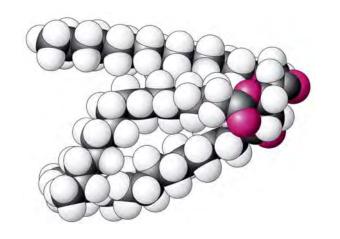
**Triacylglycerol** 

### Triacylglycerols (TAG)

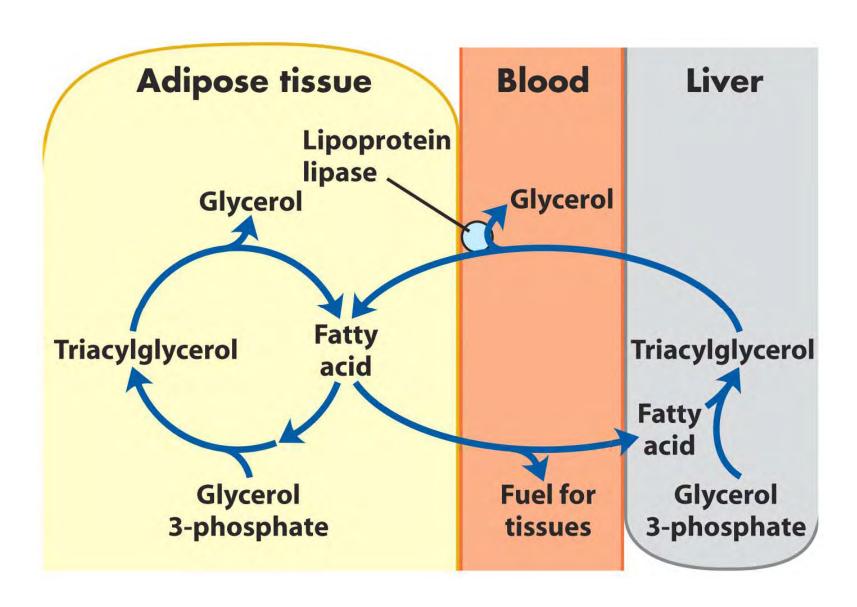
- Stored energy in fat cells
  - adipose cells



- More energy/gram than carbohydrates
  - 9 kcal/g compared to 4 kcal/g
  - stored in anhydrous form (carbs 2g H<sub>2</sub>O/g)
  - => fat has 6.75x > energy than hydrated glycogen

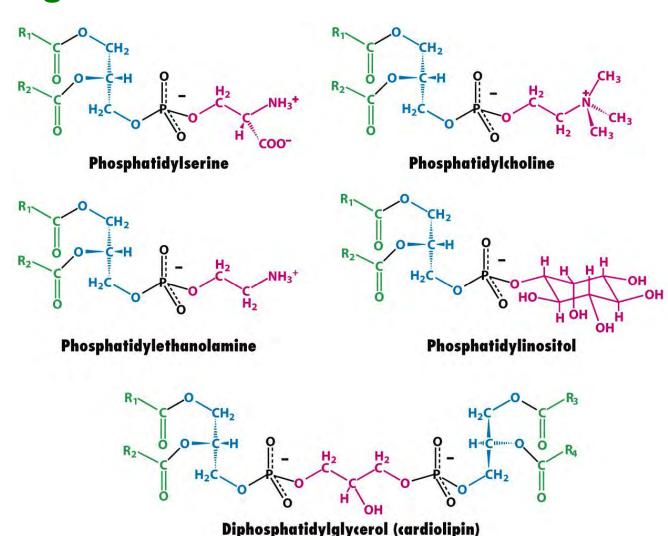


### **TAG** cycle



#### Glycerophospholipids (phosphoglycerides)

glycerol-based PLs – main component of biological membranes



	Name of glycerophospholipid	Name of X	Formula of X	Net charge (at pH 7)
PA	Phosphatidic acid	(4)	— н	- 1
PE	Phosphatidylethanolamine	Ethanolamine	- CH <sub>2</sub> -CH <sub>2</sub> -NH <sub>3</sub>	0
PC	Phosphatidylcholine	Choline	$-CH_2-CH_2-N(CH_3)_3$	0
PS	Phosphatidylserine	Serine	-CH <sub>2</sub> -CH-NH <sub>3</sub>	- 1
PG	Phosphatidylglycerol	Glycerol	— CH <sub>2</sub> —CH —CH <sub>2</sub> —OH	- 1
PIP <sub>2</sub>	Phosphatidylinositol 4,5-bisphosphate	myo-Inositol 4,5- bisphosphate	H O—P OH H OH HO O—P  1 OH HO O—P	- 4
CL	Cardiolipin	Phosphatidyl- glycerol	CH <sub>2</sub>   CHOH O   CH <sub>2</sub> O-P-O-CH <sub>2</sub>   O	- 2
			CH_O_C_R <sup>1</sup> CH <sub>2</sub> —O_C_R <sup>2</sup>	

### Biological role of PLs

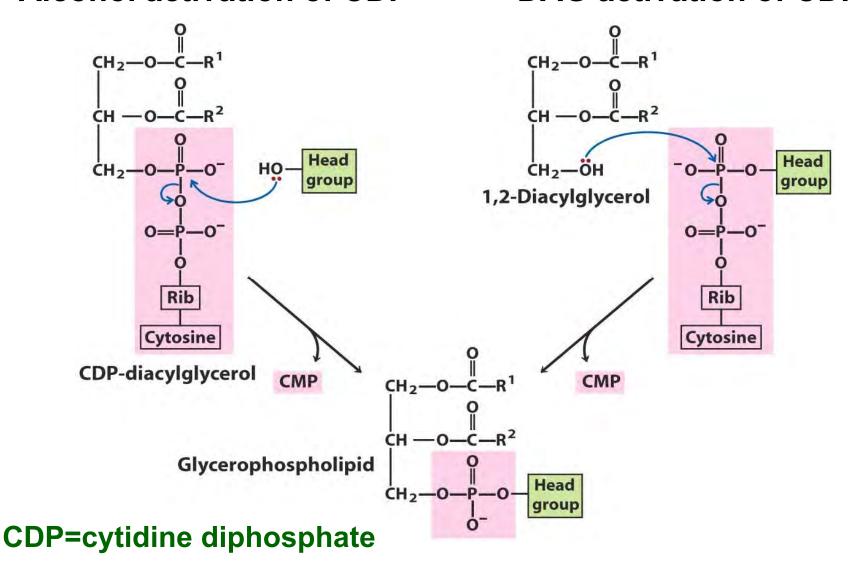
- PC most common lipid in mammals
- PS is 10% of total PLs in mammals
- CL located in inner mitochondrial membrane (role in oxidative phosphorylation)
- Ampipathic = possesses both hydrophilic and hydrophobic properties

#### PL synthesis requires an activated intermediate

Strategy 1
Alcohol activation of CDP

Strategy 2

DAG activation of CDP



$$\begin{array}{c} & & & \\ & &$$

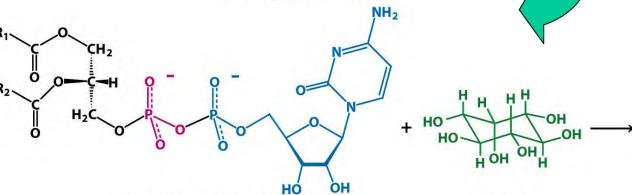
### CDP-diacylglycerol

**Phosphatidate** 

**Strategy 1** 

PL synthesis from activated alcohol

CDP=cytidine diphosphate



NH<sub>2</sub>

Inositol

OH

HO

**CDP-diacylglycerol** 

#### **Strategy 2**

PL synthesis from activated DAG

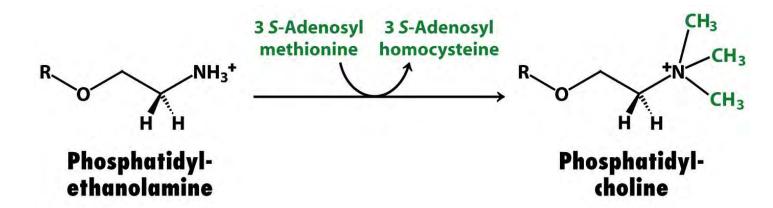
CTP=cytidine triphosphate
CDP=cytidine diphosphate
CMP=cytidine monophosphate

**Phosphatidylethanolamine** 

### Glycerophospholipid synthesis

PE and PC: Strategy 1 PI, PG, cardiolipin (CL): Strategy 2

PS, PE, PC are "coupled"
PS from PE or PC
PC from PE (+ 3 adoMet)

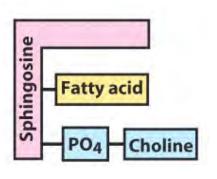


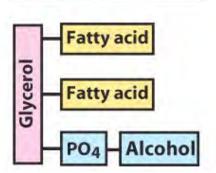
**Sphingolipids** 

contain sphingosine backbone (not glycerol)



#### **Sphingosine**

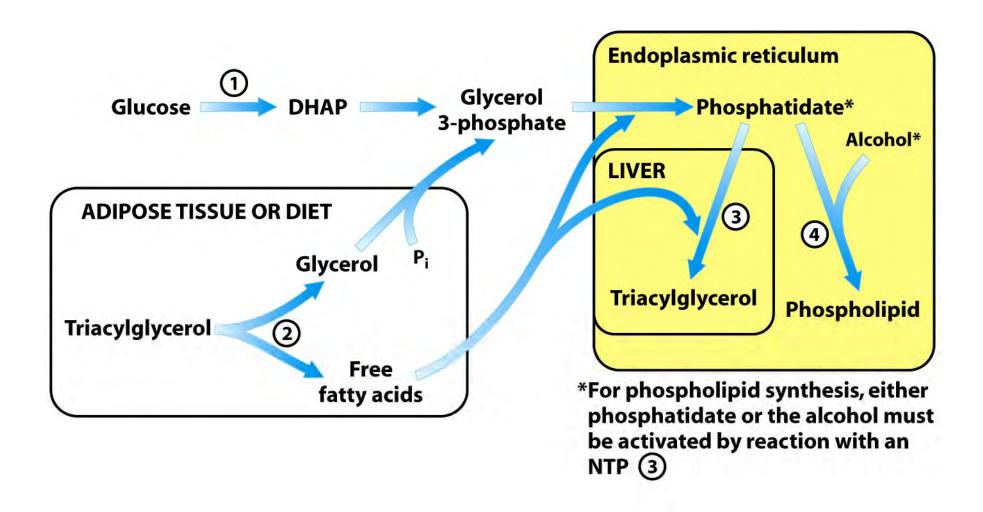




glycerophospholipid

**Phosphatidylinositol** 

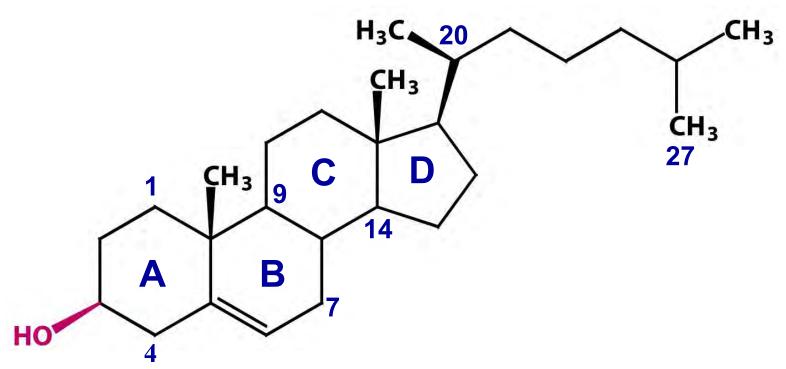
**Membrane lipid = concentration ↑ in central nervous system** 



Pathway integration for TAG and PL synthesis phosphatidate is produced from multiple pathways and is further incorporated into TAGs or PLs

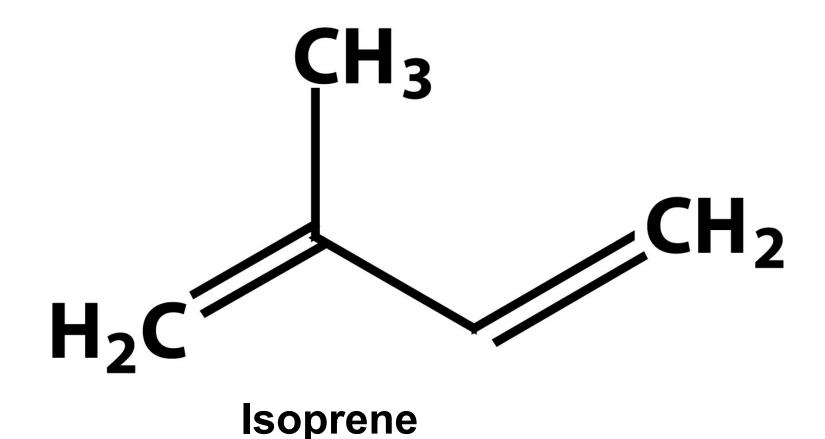
### **Cholesterol biosynthesis**

modulates fluidity in animal membranes and is precursor of steroid hormones



In 4 steps ....

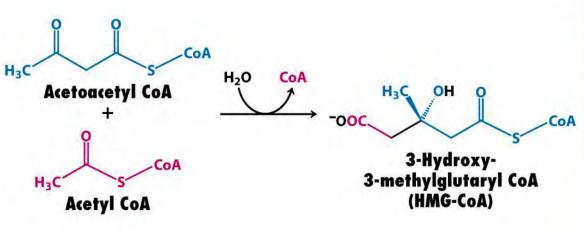
## where isoprene is the key intermediate

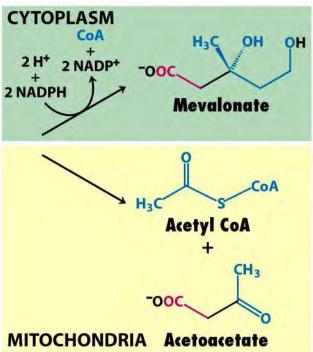


## 3 CH<sub>3</sub>—COO<sup>-</sup> Acetate CH<sub>3</sub> TOOC-CH2-C-CH2-CH2-OH Mevalonate isoprene **Activated isoprene** Squalene Cholesterol

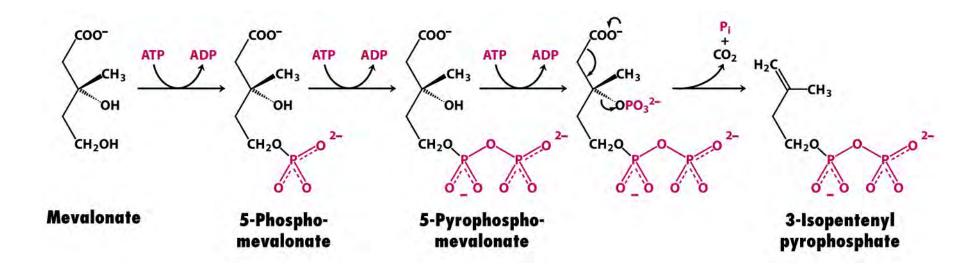
- 1. Condensation of 3 Ac-CoA to mevalonate
- 2. Conversion of mevalonate to activated isoprene (3-isopentenyl pyrophosphate)
- 3. Condensation of 6 activated isoprene units to squalene
- 4. Cyclization

## Step 1: Condensation of 3 Ac-CoA to mevalonate (6 carbons)

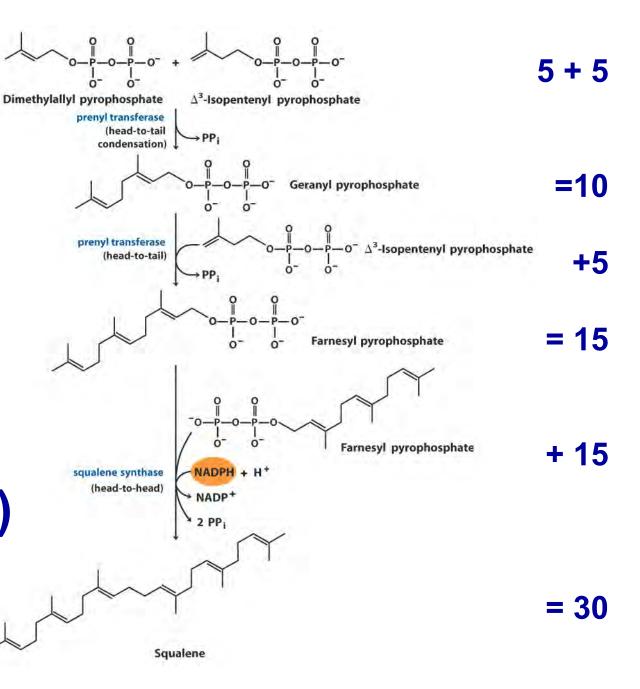




## Step 2: Conversion of mevalonate to activated isoprenes (5 carbons)

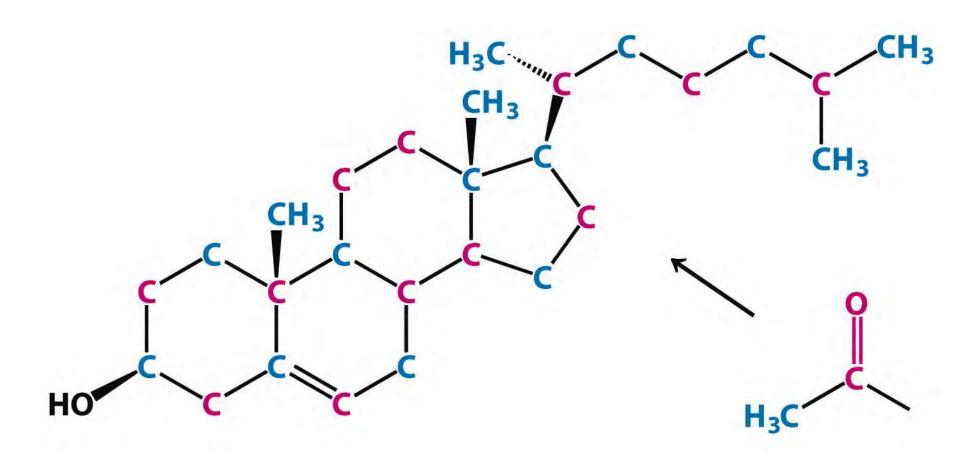


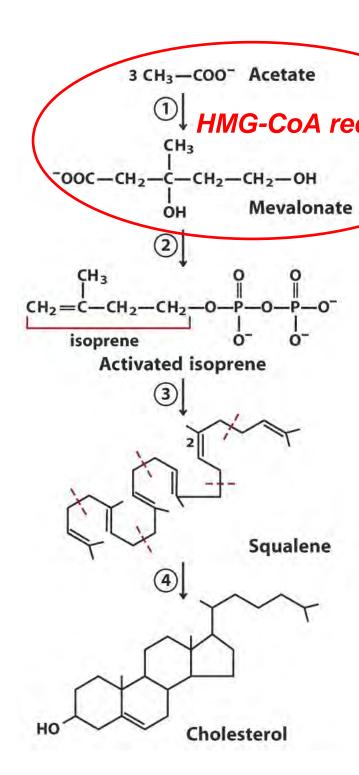
Step 3: Cond. of isopreneunits to squalene (linear, 30 carbons)



### Step 4: Cyclization (4 rings)

## All 27 carbons in cholesterol are dervied from Ac-CoA



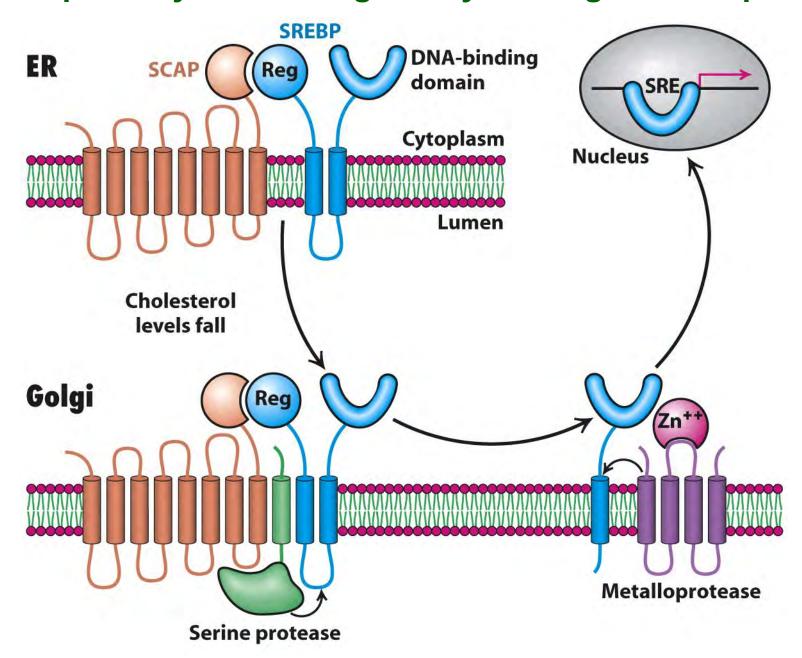


- 1. Condensation of 3 Ac-CoA to mevalonate limiting step!!
- 2. Conversion of mevalonate to activated isoprene (3-isopoentenyl pyrophosphate)
- 3. Condensation of 6 activated isoprene units to squalene
- 4. Cyclization

## Regulation of HMG-CoA-reductase (integral membrane protein in ER)

- 1. Feedback cholesterol stimulates proteolysis
- 2. Hormonal inactivated by phosphorylation, activated by dephosphorylation
- 3. Transcription via SREBP (rate of synthesis) sterol regulatory element binding protein
- 4. Therapeutics mevalonate analogs competitive inhibitor statins lovastatin, atorvastatin Lipitor

#### **SREBP** pathway = sterol regulatory binding element protein

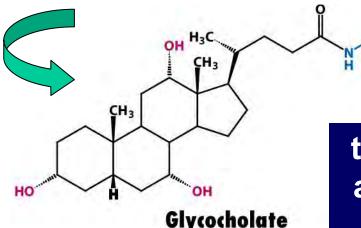


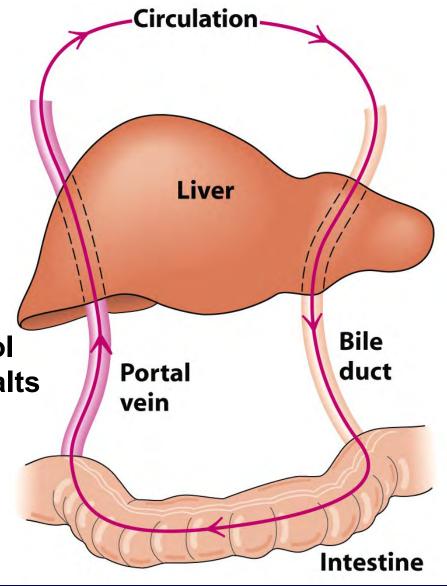
# Cholesterol is a precursor to bilesalts (and to steroid hormones and vitamin D)

## **Enterohepatic** circulation

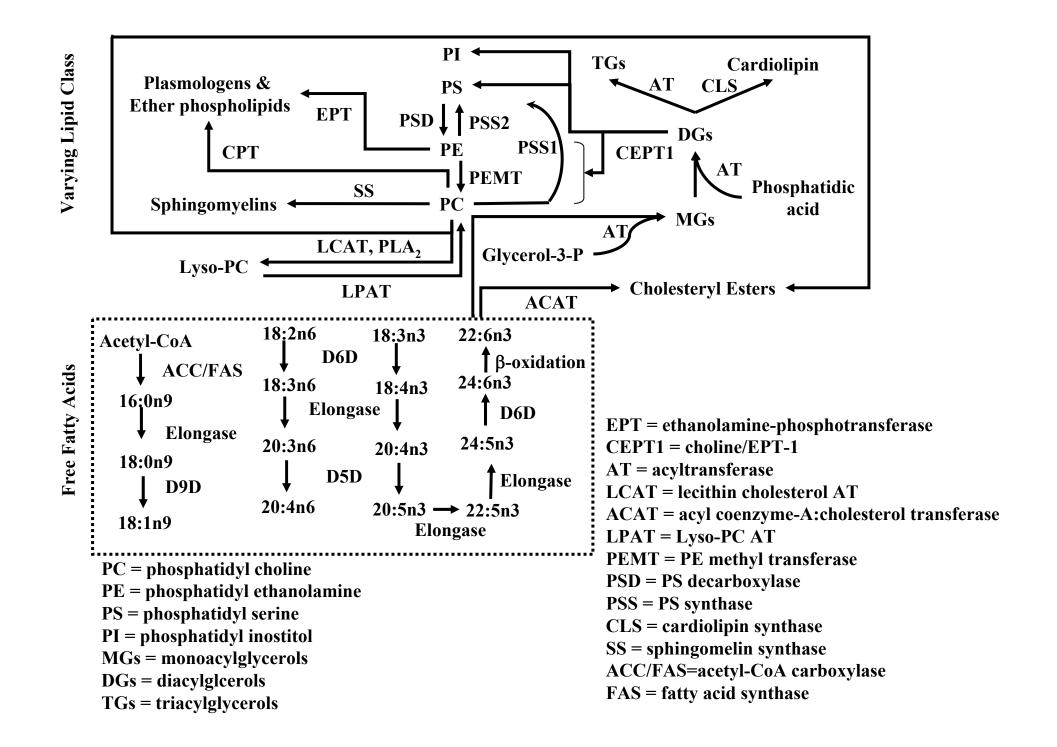
circulation of bile from the liver, to the small intestine where it aids in fat digestion

hepatocytes metabolize cholesterol to lipid-soluble bile acids → bile salts conjugated to glycine or taurine





the enterohepatic circulation of bile acids may be disrupted as a way to lower cholesterol



### What is a lipid?

#### **Classical definition:**

biological molecule that is soluble in organic solvent, but insoluble in water

#### **Modern (specific) definition:**

fatty acids and their derivatives, and substances related biosynthetically or functionally to these compounds

#### **More specific definition:**

Hydrophobic small molecules that originate by carbanionbased condensations of thioesters (fatty acids, polyketides, etc.) and/or by carbocation-based condensations of isoprene units (prenols, sterols, etc)

### Specific structural-based definition

### Lipid Biosynthesis - summary

- Lipids overview classification, structure, synthesis
- Fatty acids in liver (fat) and cyotosol; Mal-CoA; multifunctional enzyme; modified in ER
- Eicosanoids important signaling molecules; play roles in pain and inflammation
- Triacylglycerols in liver, fat, and intestine; energy storage; phosphatidate; dynamic equilibrium
- Phospholipids in almost all cells; phosphatidate, 2 strategies for synthesis, membrane components
- Cholesterol in all cells; mostly liver, 4 steps, from Ac-CoA via isoprene, HMG-CoA reductase
- Bile acids important in fat digestion, enterohepatic circulation

Whew.....



- http://www.cyberlipid.org/
- http://www.lipidlibrary.co.uk/
- http://www.lipidmaps.org/
- http://www.metabolomics.se/
  - (contains downloadable file of today's lecture under the section "Courses")

